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**“Spore-forming bacteria as indicators of pollution
in sediments of Lake Geneva”**

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Spore-forming bacteria as indicators of pollution in sediments of Lake Geneva

Short abstract:

Two components: Endospore-forming Firmicutes are well known for their capacity to survive under harsh environmental conditions through the formation of strong and persistent structures called endospores. Sediments of the Bay of Vidy (Lake Geneva), that are well known for their heavy metal pollution. The question was: Are endospore-forming Firmicutes predominant members of the bacterial communities in the most impacted sediments? The co-occurrence of these two components was effectively observed, and four particular strains of Clostridia were identified as dominant. However, their presence could be related to a combination of a dysfunction of the waste water treatment plant infrastructure and their survival in the state of an endospore. Episodic flooding events in the waste water treatment plant is effectively strongly suspected and supported by the data collected during this thesis. At the end of this work, the dominance of this particular group of bacteria does not seem to be uniquely related to their resistance to high concentrations of heavy metals, but also to the environmental changes occurring during the waste water treatment process and/or during the burial of bacterial communities in sediments.

Keywords: Endospore-forming Firmicutes, heavy metals pollution, waste water treatment plant, Lake Geneva sediments

PREFACE

This thesis was submitted to the Faculty of Science at the University of Neuchâtel (Switzerland), as a partial fulfilment of the requirements to obtain a degree of PhD. The work presented was carried out during three years, from 2011 to 2014, in the laboratory of microbiology headed by the assistant professor Pilar Junier. These three years were however not entirely dedicated to thesis work, but also to very interesting, but time consuming, academic duties with bachelor and master students. As a counterpart, the collaboration with a master student (as Master thesis supervisor) during the first year of the project was very helpful and stimulating.

The title of this thesis work is "Spore-forming bacteria as indicators of pollution in sediments of Lake Geneva". However, with the different findings and the inevitable re-orientation that occurred all along such a work, it should perhaps have been more correct to re-entitle it as follows: "Co-occurrence of trace metals and endospore-forming Firmicutes in sediments of Lake Geneva contaminated by human faeces". The project is part of the Elemo (exploration des eaux du Léman) project initiated thanks to Ferring Pharmaceutical and the EPFL (Ecole polytechnique fédérale de Lausanne). By grouping together scientists from different fields, but also from different Swiss and foreign universities, the Elemo project aimed at a better understanding of the bio-, geochemical functioning of Lake Geneva. Because of the increase of the population around its shore, Lake Geneva experienced in the recent decades an ever increasing anthropic pressure, which needs to be mastered to preserve the quality of its water. One and a half million people live on Lake Geneva's shore and about one half of them depend on it for drinking water resources. Furthermore, Lake Geneva is part of the Rhone river, which provides different water services for an important part of France. Preserving the quality of this water resource is therefore capital for the future.

During summer 2011, thanks to the collaboration with the "Consulat honoraire de la fédération de Russie à Lausanne", two Russian MIR submersibles were available for scientific field work in Lake Geneva. The presence of the submersibles in the lake was a large media event for EPFL and Ferring Pharmaceutical, but at the same time it provided to the scientists an incredible opportunity to examine more in detail different aspects of the water column, as well as the sediment floor of the lake.

As different scientists from different disciplines were involved in this project it was also, for each of us, an incredible opportunity to collaborate and to learn things from the others. This thesis project would certainly not have been the same without the helpful collaboration we had with people from the University of Geneva.

As said by Philippe Gillet (Vice President for Academic Affairs at the EPFL) during an interview, the Elemo project is a beautiful means to meet and to learn between different cultures, but also between different scientists.

Why is it so important to preserve the environment?

We often hear environmentalists saying that we should take more care of the environment, by preserving species and biodiversity. In most of the cases, I however think that people understand these recommendations as a means to preserve something that is not really useful for humans.

Many people see nature as a place, outside the cities, where some farmers are working to produce what people eat. For others, nature is a place where one can make some leisure activities or see beautiful animals or plants. For all of them, nature is something our species can exploit without limit, to meet our needs, as a kind of service specially dedicated to humans. This particular vision of the world is in fact probably due to the central place occupied by humans in the Christian religion. Effectively, as children of God, nature is perceived more as a theater set rather than the cradle of every life form on Earth.

I, however, think that the main problem of this point of view, is that people do not consider themselves as a part of this complex ecosystem, and therefore do not understand why it is so important to preserve the natural environment. They have not yet realized, that if nature is impaired, the human species will also be impaired, in a way or another. In their book, "What's life?", Lynn Margulis and Dorion Sagan gave an interesting but unusual definition of humans: "People, for example, redistribute and concentrate oxygen, hydrogen, nitrogen, carbon, sulfur, phosphorus, and other elements of Earth's crust into two-legged, upright forms that have an amazing propensity to wander across, dig into, and in countless other ways alter Earth's surface. We are walking, talking minerals!"¹

In an evolutionary perspective, our species is on the top of the tree of life, far away from its trunk. One can see this position as an advantage giving us a kind of superiority on other "primitive" species. One must however not forget that this position is also the result of an extremely fragile equilibrium that depends on processes occurring thanks to organisms at the base of the tree of life. The life of "superior" organisms is effectively closely linked to the services provided by microbial communities in the ecosystems. A better understanding of their function is, therefore, necessary to understand how ecosystems work and how one can preserve them against deleterious human activities. Preserving the planet will first benefit ourselves! That is why research projects in environmental science are so important.

THESIS OBJECTIVES

The global objective of this thesis was to achieve a better understanding of the relationship between endospore-forming Firmicutes and heavy and trace metal pollution accumulated in freshwater lake sediments. The thin red line followed all along this thesis could probably

be summarized as: "How to find a way to get the least biased vision of a particular portion of the bacterial community inhabiting lake sediments ?" If this condition seemed relatively easy to achieve in a first approach, this was, however, not always the case as the work progressed. During the three years at the University of Neuchâtel, my knowledge of this particular subject has effectively evolved, highlighting, step by step, different biases. Bypassing them was always the result of choices aiming at a minimization of the distortion of the reality. But what is exactly reality when we are talking about microbial communities? It's hard to get an unequivocal answer to this question!

In summary, if this awareness of the limits of science has greatly contributed to my personal education, I also hope that the different methodological and computational biases that were identified all along this thesis, as well as the way they were bypassed, will contribute, in the future, to an increase of our knowledge on endospore-forming Firmicutes, but also on the fate of heavy and trace metals in the environment.

THESIS OUTLINE

This thesis is in the form of a synopsis with attached publications. It is divided into three parts. Roughly, Part I is the core of the thesis and is composed of global informations on the subject, published papers and finally a conclusion. Part II is constituted by additional publications in which the author has contributed. Finally, part III is composed by an updated *Curriculum Vitae*.

In Part I, the first chapter introduces rapidly some particular properties of water. It gives an overview of the environmental issues linked to the pollution of fresh water resources with a particular emphasis on heavy metals collected by run-off water. The main remediation techniques used in heavy metal polluted soils and sediments are finally exposed. In a second part of chapter one, some basic information on bacteria is provided. This part defines, in particular, the kind of bacteria found in lake sediments, but also in waste-water treatment plants. This second part of chapter one ends with general informations on endospore-forming bacteria, as well as with a small description of the bacterial resistance mechanism for fighting against high environmental concentration of metals. Finally, in a third section of chapter one, a global presentation of Lake Geneva and the pollution issues in the Bay of Vidy is provided, with a particular emphasis on the rejection of the effluent of the waste water treatment plant of Lausanne. The functioning of the plant is then briefly described and this section ends with a small description of separation status of waste and run-off water for the city of Lausanne.

Chapter 2 tries to give a global overview of key methodological particularities used all along this work. The chapter starts with the description of the sampling procedure with MIR submersibles and the research boat from the Forel-Institute (University of Geneva, Switzerland). A quick description of the different methods available to study microbial communities is then provided. This chapter reports also the preliminary experiments performed to select and design the more efficient method to extract DNA from endospore-forming Firmicutes buried in Lake sediments. Finally, it describes also the

pyrosequencing of the DNA samples and the bioinformatic treatment of the sequences. The chapter concludes with a quick description of the chemical analysis performed in this thesis.

Chapter 3 is composed of the publications of the author. It starts with one conference proceeding, "Endospore-forming bacteria as an indicator of pollution in sediments of Lake Geneva", reporting the participation to ICHMET 2012 (International conference on heavy metal in the environment), in Roma. This is followed by the *in press* article in Aquatic Sciences, "Bacterial communities in trace metal contaminated lake sediments are dominated by endospore-forming bacteria"; the published article in Applied and Environmental Microbiology, "Quantification of endospore-forming Firmicutes by quantitative PCR with the functional gene *spo0A*"; and finally, the chapter ends with the article in preparation for Environmental Science and Technology, "Endosporulation as a resistance mechanism to metal-toxicity in sediments"

Chapter 4 describes laboratory experiments performed with four Clostridia strains and different heavy metals. Initially, it was planned to publish the results presented in this chapter. Experiments were however not sufficiently accomplished to allow this. These experiments explored the effect of heavy metal on sporulation rate and growth dynamic of the four Clostridia strains. During these experiments it was observed that heavy metals precipitated readily in the culture vials. Experiments to try to understand if this precipitation phenomena is related to physico-chemical conditions prevailing in the vials or to an immobilisation phenomena by bacterial biomass are finally reported.

Finally, Chapter 5 is composed by a synthesis of the work performed during this thesis, giving the main conclusions that can be drawn, as well as their outlook. A critical review of the different bias encountered all around this thesis is also provided in this chapter.

LIST OF PAPERS

- I Bueche, M., Sauvain, L., Wunderlin, T., Junier, T., Tercier, M., Loizeau, J.-L., Masson, M., Gascon, E., Junier, P.
Endospore-forming bacteria as an indicator of pollution in sediments of Lake Geneva
E3S Web of Conferences 1 (**2013**),doi: 10.1051/e3conf/20130133011
- II Sauvain, L., Bueche, M., Junier, T., Masson, M., Wunderlin, T., Kohler-Milleret, R., Gascon D., E., Loizeau, J.-L., Tercier-W., M.-L., Junier, P.
Bacterial communities in trace metal contaminated lake sediments are dominated by endospore-forming bacteria
Aquat. Sci. (**2013**), doi: 10.1007/s00027-013-0313-8
- III Bueche, M., Wunderlin, T., Roussel-Delif, L., Junier, T., Sauvain, L., Jeanneret, N., Junier, P.
Quantification of Endospore-Forming Firmicutes by Quantitative PCR with the Functional Gene spo0A
Appl. Environ. Microbiol. (**2013**), doi: 10.1128/AEM.01376-13.
- IV Bueche, M., Sauvain, L., Junier, P.
Endosporulation as a resistance mechanism to metal-toxicity in sediments
In preparation... To be published in Environmental Science and Technology.
- V Wunderlin, T., Corella, J. P., Junier, T., Bueche, M., Loizeau, J.-L., Girardclos, S., Junier, P.
Endospore-forming bacteria as new proxies to assess impact of eutrophication in Lake Geneva (Switzerland–France)
Aquat. Sci. (**2013**), doi: 10.1007/s00027-013-0329-0
- VI Wunderlin, T., Junier, T., Bueche, M., Corella, J. P., Loizeau, J.-L., Girardclos, S., Junier, P.
Endospores in sediment as a proxy for 20th century environmental lake history
In preparation... To be published in Nature communications.
- VII Filippidou, S., Bueche, M., Wunderlin, T., Junier, T., Sauvain, L., Roussel- Delif, L., Jeanneret, N., Dorador, C., Molina, V., Ioannidou, A., Vargemezis, G., Johnson, D., Junier, P.
Survival strategy meets classic ecological theories: the case of endospore-forming bacteria diversity and abundance in extreme environments
Abstract for ISME2014. To be published in ISME journal.

CONTRIBUTION REPORT

Papers I, II were written in close collaboration with Loïc Sauvain, a master student in our laboratory who has chosen to do his master work on this subject. In these two publications, I performed 75% of the design of the experiments, 50% of the sampling procedure, 50% of the fluorescence microscopy and cell counting, 50% of the DNA extraction, and subsequent molecular characterisation of the bacterial communities, 100% the bioinformatic analysis, 75% of the statistical analysis, and, finally, 75% of the data representation.

Paper III was written by myself. In this publication, I performed 75% of the evaluation of target genes involved in endospore formation, 100% of the elaboration of a refined database of endospore-forming Firmicutes sequences, 100% of the primer development, 100% of the *in silico* specificity test by BLAST, 75% of the strains cultivation, 75% of the experimental accuracy tests, 100% of tuning of the PCR and qPCR conditions, 100% of the PCR and qPCR experiments with pure strains, 50% of the characterization of the environmental sequences, and, finally, 100% of the data analysis and representation.

Paper IV was written 75% by myself. In this publication, I performed 75% of the sampling, 50% of the chemical analysis, 75% of the DNA extraction, 100% of the bioinformatic analysis, 75% of the qPCR on 16S rRNA and *spo0A* genes, 100% of the DPA quantification, and, finally, 100% of the data analysis and representation.

Paper V and VI are the main work of T. Wunderlin where I contributed with the development of the DNA extraction methodology and the development of the qPCR quantification of endospore-forming bacteria. Finally, the seventh article, which is in fact not yet published, is the main work of S. Filippidou where I contributed with some DNA samples, as well as methodological, bioinformatic, and statistical analysis issues.

The first four papers are presented in the core of the thesis, whereas the fifth to the seventh are presented at the end.

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Part I

CORE OF THE THESIS

BACKGROUND AND MOTIVATION

The aim of this first chapter is to highlight the motivation of the research conducted in this thesis. It begins with an introduction to aquatic environments and the pollution problems they are facing under anthropic pressure. Then an overview of the bacterial world is presented with a particular focus on endospore-forming bacteria. Finally, a short description of Lake Geneva is added to introduce the sampling environment used in this study.

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1.1 Water

Water is an essential constituent of the biosphere in which it is present in each of its three phases: gaseous, liquid, and solid. Water is the main substrate for biological activities and also the main constituent of living organisms (up to 70% of their weight). The quantity of water on earth is estimated at around $1350 \times 10^6 \text{ km}^3$. 97% of this quantity is represented by the oceans and the remaining 3% is distributed between soft continental water (0.6%) and ice and snow mainly from polar caps (2.4%)(Figure 1.1). A very small fraction is also contained in the atmosphere (as clouds and vapours) and in living organisms (around $1 \times 10^{-4}\%$ of the total).²

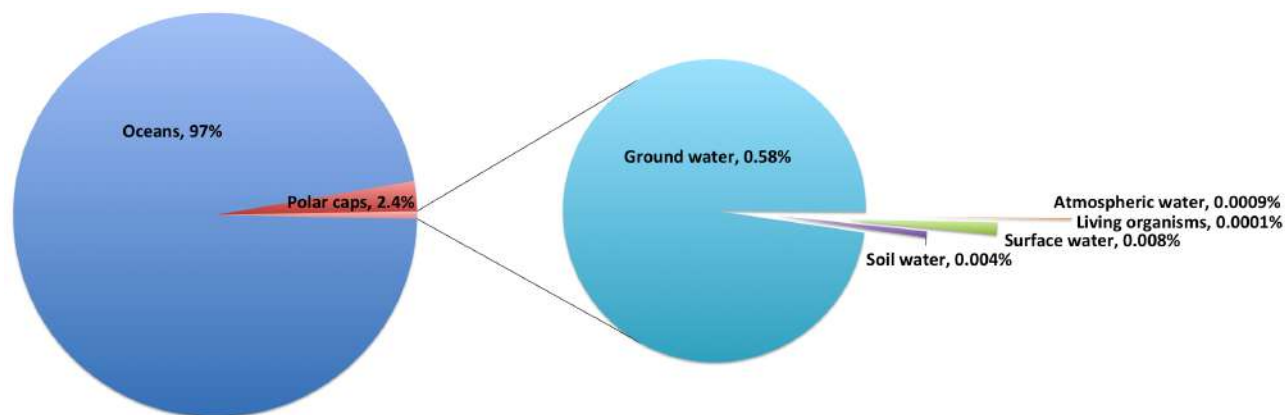


Figure 1.1: Repartition of water resources on earth.

Thanks to its physical and chemical properties, water plays a predominant role on climate regulation on the Earth. As universal solvent it allows the availability and the transport of nutrients in ecosystems.³ Through dissolution and precipitation phenomena, water provides the main link between the geological cycles of elements from rocks to the uptake and temporal sequestration of them by the biosphere. This is the "coupling belt" between slow geological and fast biological cycling of elements on Earth.

Particular properties of water. As for other chemical compounds, molecules of water become closer when temperature decreases. This is, however, only true down to 4°C where fresh water reaches its maximum density. At a lower temperature fresh water starts to freeze and molecules become more distant from each other. This phenomenon is called water thermal anomaly and has consequences on the functioning of aquatic ecosystems. For instance, this mechanism allows water stratification in lakes and oceans, leading to particular dispersion of nutrients and dissolved gases in the water column during different periods of the year. The fact that ice density is lower than the density of cold water is also interesting because it allows ice to float at the surface of water bodies. This ice sheet isolates the underlying water column from a cold atmosphere and allows living organisms to stay alive during the coldest seasons (Figure 1.2).

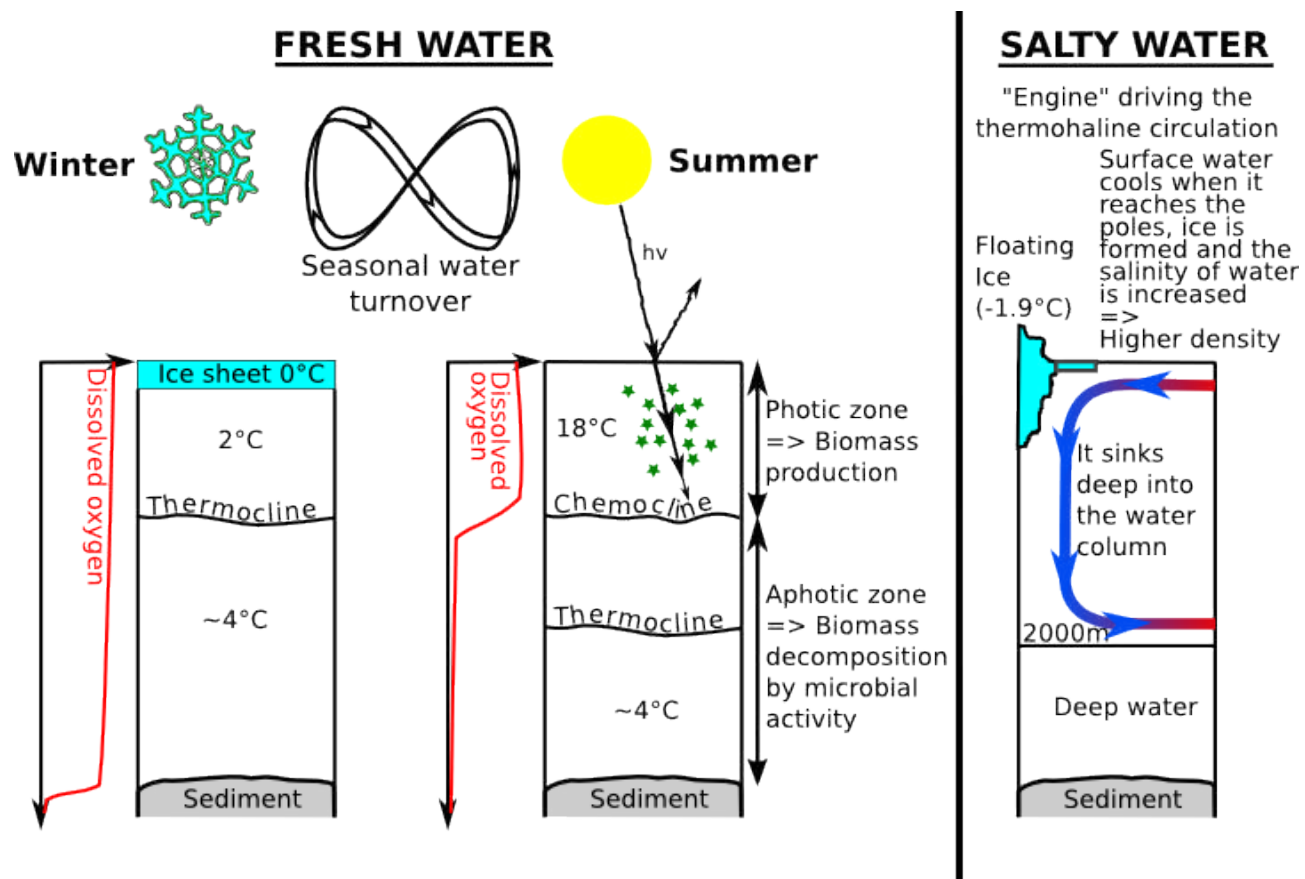


Figure 1.2: Illustration and implication of the water thermal anomaly in fresh and salty water.

The atoms in the water molecule are asymmetrically positioned giving to it polarity (negatively charged in the oxygen side, positively charged in the hydrogen side). This polarity allows the formation of hydrogen bonds between water molecules leading to some of its more interesting properties. In physics, the amount of energy needed to change a compound from one state to another (for instance from liquid to gaseous) is called latent heat. For water, latent heat is exceptionally high (only ammonia has higher values) and plays a capital role in the regulation of temperature on Earth. The presence of liquid water on Earth also derives from this high capacity to absorb energy without changing state. The partial charge of the water molecule also explains why water is so efficient as a solvent for salts (ions) or other polar compounds. The dissolution of salts also has an influence on the properties of water. For instance, salty oceanic water, in contrast to fresh water, has a maximum density at a temperature below its freezing temperature. In the polar zones, where surface water is strongly cooled, this allows a large proportion of it to descend deep in the water column before the formation of ice. This phenomena is part of the principal engine driving the well known thermohaline circulation, which determines the oceanic currents all around the world. Atmospheric currents and the different climatic zones are greatly determined in response to this oceanic circulation.⁴

In the atmosphere, water vapour absorbs infra-red radiations emitted by Earth on reaction to solar irradiation. By itself this absorption accounts for about two thirds of the natural greenhouse effect. The formation of clouds and raining events are also a derivative from the polarity of water molecules. This comes from the high surface tension of water which allows for the formation of drops in the clouds.

From a biological perspective, the specific properties of water have also played a

glucose) by the cellular machinery. In nature, organic carbon is found in large quantities in polymers, such as starch, fatty-acids or proteins. The first step needed to use energy from these complex molecules is a hydrolysis (the addition of a water molecule) for the break-down into monomers. These monomers, as explained above can either be used to produce energy, or can be assembled to build new polymers through anabolic processes. Polymerisation occurs via a dehydration step. Because life machinery is so complex, and because water is at the center of the main metabolic reactions, the need of high quality pure water is, therefore, essential for sustaining life on Earth. But what does exactly mean HIGH QUALITY PURE water?

1.1.1 Pollution

What is exactly pollution? Is it a measure of harmfulness for living organisms? In fact the answer to these questions is not easy to find. In nature a lot of substances, like cyanides or mycotoxins (like aflatoxins, for instance)⁷ have harmful effects on life, but we do not normally consider them as pollution. The reason for this is that these substances originate from nature. But in this case, what about the poisoning of lakes by microcystins produced by some species of cyanobacteria?⁸ Of course, their success is often the result of an eutrophication phenomenon; but microcystin-producing cyanobacteria have not been introduced in nature by humans. The notion of pollution has in fact greatly evolved during the last century in response to the increasing demography, the intensification of land-use, the industrial development, the progress in environmental sciences, and also in measuring technologies.⁹ As a summary one can say that pollution occur when a substance with anthropic origin accumulates somewhere and generates a harmful effect directly on humans or on an ecosystem that provides a service to humans. Pollution can also be generated by the proliferation of a particular harmful organism under anthropic pressure. This focus on human benefits is important because it explains how the notion of pollution has evolved during the last decades. Ecology is the discipline of life sciences which tries to understand links between organisms and ecosystems. Notable advances occurred in this discipline thanks to the systems theory. To understand that almost all ecosystems are linked together it was effectively first necessary to understand natural environments in a systemic point of view. With such a comprehension of nature one can easily understand that every source of pollution can affect human health in a way or another through time.

Sources of pollution. What are the main sources of pollution known today? Pollution can be divided into two main categories depending on the way they enter natural ecosystems. Point-source pollution correspond to a large amount of a pollutant that is released in a particular point. Landfills and industrial effluents are two examples of this. In developed countries this kind of pollution source is generally under control because infrastructures to avoid pollutants dispersion are imposed by law.¹⁰ In Switzerland, for instance, domestic as

sediments, even at a pH as low as 4 or 4.5. This suggests that heavy metals are bound in sediment by sorption, and by the precipitation of secondary minerals.¹⁵ In natural environments conditions are, however, not as simple as in the laboratory. The chemical speciation of metals is influenced by many environmental factors, like salinity, sediments granulometry, hardness (pH buffering effect of calcite), Eh, and pH conditions. All these factors could finally have an impact on the metal binding and release, as well as their bio-availability and toxicity.¹⁶ Microbial activity in sediments often consumes very rapidly dissolved available oxygen, leading to a decrease of redox potential. In such conditions, when sulphates are present, sulfides are produced and can immobilise divalent metal cations through precipitation of secondary metal sulfides. These minerals are stable as long as the redox potential is maintained at a low value. Aeration, pH decrease or organic matter mineralization could lead to a release of these temporarily trapped heavy metals in the environment.¹⁷

All around the world sediment pollution by heavy metals is reported by scientists, particularly in areas with the higher densities of population or particular human activities, like harbours or ore extraction sites. A search in web of science for titles containing "heavy metal" and "sediment" as keywords reports close to 500 publications in the last fifteen years, giving us a global idea of the extend of the issue of sediments pollution by heavy metals. In Switzerland, for instance, values higher than national guidelines on soils contamination are regularly reported for lake or river sediments sampled downstream from major urban areas.¹⁸ In the last years, ICPR (International commission for the protection of Rhine) has defined some target values for the sediments of Rhine and has edicted rules to reduce inputs of heavy metals in the near future. The values retained were the following: 1 *mg/kg* for Cd, 50 *mg/kg* for Cu and Ni, 0.5 *mg/kg* for Hg, 100 *mg/kg* for Pb and 200 *mg/kg* for Zn.¹⁹

Plants are well known for their capacities to concentrate toxic metals in their tissues.¹⁴ In many places of the world, heavy metal rich waste water is used to irrigate cultures, leading to an accumulation of these metals in fruit and vegetables.²⁰ Similarly, an accumulation of heavy metals is also reported in vegetables irrigated by polluted water²¹ or grown on polluted soils. Cultures grown in Romania in heavily polluted sites have reported concentrations of Cd and Pb in fresh root vegetables that exceed 1.25 times and 11 times, respectively, the limit fixed by the world health organisation (WHO). In leafy vegetables contents of heavy metals were 7 times higher than normal concentrations for Cd and 17 times higher for Pb.²²

of heavy metal is minimal and can only be efficient through mutation.⁶³ Once inside the cells, heavy metals can easily link to *SH* groups in proteins, strongly perturbing enzymatic activities. Some heavy metal could also compete with other essential metallic ions needed to perform metabolic activities.²⁴ For instance, Cd^{2+} could replace Zn^{2+} in DNA-binding proteins (like zinc-finger proteins), or Ca^{2+} , Ni^{2+} and Co^{2+} could replace Fe^{2+} in heme proteins or in different siderophores. Finally, elevated concentrations of non-essential heavy metals generate oxydative stress inside the cells.²⁴ The solution to regulate an excess of undesirable intracellular heavy metals is to use "metal-ion homeostasis factors" and "metal-resistance determinants".⁶³ As heavy metals cannot be degraded, three main solutions are available to cope with these intracellular excesses.

1. Active efflux of specific ion (via specific inducible transporters)
2. Segregation / complexation (immobilisation) of undesirable metallic cations into complex compounds containing thiol-groups. This solution, however, requires large quantities of energy (ATP) to synthesize sequestration proteins.
3. Diminution of the toxicity by reducing heavy metals into a less toxic form (reduction of Hg^{2+} to Hg^0 , for instance).

Active efflux mechanisms can be mediated by P-type ATPases (for Mg^{2+} , Mn^{2+} , Ca^{2+} , K^+ , Cu^{2+} , Zn^{2+} , Cd^{2+} , Pb^{2+} , and Ag^+), A-Type ATPases (arsenite), proton antiporters (Co^{2+} , Zn^{2+} , Cd^{2+} , Ni^{2+} , Cu^{2+} , and Ag^+), or by specific slow chemiosmotic transporters (Zn^{2+} , Cd^{2+} , Co^{2+} , and Fe^{2+}). Active influx could also be mediated by ABC-binding cassettes (Mn^{2+} , Zn^{2+} , Ni^{2+} , and Fe^{2+}). This last transport system is, however, not active for the efflux of these metallic ions.⁶³

Extracellular complexation mechanisms could also play an important role in the bio-availability of heavy metal ions. Complexation by exopolymeric substances,⁶⁴ or S-layers^{65,66} is effectively reported by many authors. This extracellular immobilisation system is, however, not specific, and lead also to a diminution of the bio-availability of essential metals and nutrients.

Finally, the last mechanism to escape elevated concentrations of heavy metals could be endosporulation. This is the purpose of this work!

1.3.3 Runoff water of Lausanne

The separation of runoff and waste water is not yet completely achieved in Lausanne and an amount of clear water is still going through the waste water treatment plant of Vidy. This input of clear water in the waste water treatment plant is not desirable for two reasons. First, the induced dilution of waste water reduces the efficiency of the treatment. Second, it is an energetic, economic, and ecological nonsense to sully this surface runoff water with faecal contaminants to treat it thereafter in installations which are not designed for this kind of water. In few years the city of Lausanne has invest a lot to separate clear water from waste water. Storm overflow of "Le Capelard", operated in 2012 is a good example of this. Now a big proportion of clear water is discharged directly into the lake.

But what about heavy metals contained in this runoff water? Of course, they are probably not those we found associated with the sediments sampled around the rejection pipe, but they still induce pollution of the aquatic ecosystem of the lake! What is, therefore, their long term impact on the health of the ecosystem and our health? Nobody knows! A precautionary approach should dictate to us to avoid this kind of rejection into the environment and to found a way to eliminate heavy metal pollution from clear water before their rejection in nature. This research project should gives some clues for solving this problem.

2

CONCEPTS AND METHODS

This chapter introduces some key aspects of the theoretical, experimental, and computational concepts used throughout the thesis

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During MIR sampling campaign (summer 2011), different aspects of sediment surface were reported depending of the sampling zone (Figure 2.4). Close to the rejection pipe sediments were covered by a white layer of filamentous bacteria that look like *Beggiatoa* sp. When coring in this area, a great amount of gas bubbles were also emitted by sediments, reporting an intense mineralisation of a high load of organic matter in anaerobic conditions. Gas produced could be a proxy for methanogenesis (CH_4), as well as sulphate reduction (H_2S), fermentation (H_2) or nitrate reduction (N_2). All these process could take place in anoxic sediments in different micro-niches depending on the local availability of the substrates as well as the local redox conditions at a microscale. However, H_2S is highly soluble, H_2 is consumed by different metabolisms (like methanogenesis), and the low concentrations of nitrate in this aquatic environment⁷ does not allow the formation of large quantities of N_2 . At the end, only methane (CH_4) can be degassed by sediments.

